

Ex vivo* Anti-Oxidation Activity of Polysaccharides from Red Alga *Porphyra yezoensis

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Abstract: In this study, the *ex vivo* anti-oxidation activity of polysaccharide extracted from the red seaweed *Porphyra yezoensis* was studied. The scavenging efficiencies for free radical O^{2-} , $\bullet OH$ and $\bullet DPPH$, the inhibitory effects on mice erythrocytes hemolysis induced by H_2O_2 as well as on the lipid peroxidation in mice liver homogenate, were systematically measured and analyzed. Results suggested that *Porphyra* polysaccharide, in the experimental concentration range, possessed anti-oxidation activity. The scavenging efficiency for O^{2-} was found to be remarkably high and the maximum scavenging rate was 82.77%. The Scavenging efficiency for $\bullet OH$ was even higher than O^{2-} with a maximum removing rate of 85.63% and the scavenging for $\bullet DPPH$ was 13.97%. The inhibitory effects on mice erythrocytes hemolysis and MDA formation in mice liver were significant and the maximum inhibition rates were 82.90 and 58.48%, respectively. The combined data indicated that the polysaccharide extracted from *Porphyra* had strong anti-oxidation activity.

Key words: *Porphyra yezoensis*, polysaccharide, anti-oxidation, free radical, DPPH, MDA

INTRODUCTION

Oxygen-derived free radicals are common byproducts in the metabolism process. However, these compounds are highly active and can cause apparent damages, commonly known as the oxidative damages, to the cell membranes and other cellular structures (Cuzzocrea *et al.*, 2001). The free-radical damage has been reported to directly associate with diseases such as atherosclerosis, cataract formation, ageing and carcinogenesis (Cuzzocrea *et al.*, 2001). To counteract the destructiveness of the free radicals, human body possesses a complex antioxidant defense system that utilizes various vitamins, minerals and other naturally producing substances (Cuzzocrea *et al.*, 2001). It is possible, therefore, to prevent the radical damage by supplementing the diet with certain food, nutrients and herbs that have antioxidant activity.

In recent years, sulfated polysaccharides from the marine brown algae, *Fucus vesiculosus*, have been demonstrated to have antioxidant activities (Ruperez *et al.*, 2002). Xue *et al.* (2001) also confirmed this activity by the antioxidant functional assay through ferric reduction. Sulfated polysaccharides from *Laminaria japonica* and *Ecklonia kurome* Okam have also been

reported to possess free radical scavenging activities (Ruperez *et al.*, 2002; Hu *et al.*, 2001; Han *et al.*, 2006). More recently, data from various research laboratories have suggested that the polysaccharide from the red alga, *Porphyra haitanensis* or *P. yezoensis*, had multiple biological functions, such as anti-coagulant, anti-senescence (Zhang *et al.*, 2002), anti-fatigue (Guo *et al.*, 2005), anti-cancer (Zhang *et al.*, 2001), anti-virus (Zhou and Chen, 1990). Effects, proliferation promotion effects for lymphocytes and sertoli cells (Xiao *et al.*, 2003; Guo *et al.*, 2006), immune-modulation and free radical scavenging (Zhou and Chen, 1989; Yashizawa *et al.*, 1993).

The red alga, *Porphyra*, has been an important food source in the Asian diet, as well as a drug used in the traditional Chinese medicine. Previously, three polysaccharide fractions with different sulfate content from *P. haitanensis* have been identified to have anti-oxidation activity (Wang *et al.*, 2004) and two polysaccharide fractions from *P. yezoensis* have been discovered to possess anti-fatigue, enhancement of immunity and anti-cancer properties (Guo *et al.*, 2005, 2006; Zhang *et al.*, 2007). In this study, the anti-oxidation activity of polysaccharides from *Porphyra yezoensis* was studied and the mechanism of the polysaccharide bioactivity from *Porphyra yezoensis* was proposed.

MATERIALS AND METHODS

Materials: The polysaccharides, extracted from *Porphyra yezoensis*, were provided by the Lab of Seaweed Biotechnology, Shanghai Fisheries University, Shanghai, China. They were purified by DEAE-52 and Sephadex G-200 columns following the protocol as previously described (Liu *et al.*, 2005; Gu *et al.*, 2007).

Measurement of the superoxide radical scavenging activity: The method for the measurement of superoxide radical scavenging activity is similar to the protocol reported previously (Ponti *et al.*, 1978). Briefly, superoxide radicals were generated by the PMS-NADH-NBT system. The reaction solution (3 mL) contained 0.5 M Tris-HCl (pH 8.0), 80 μ M deoxy-coenzyme I (NADH), 50 μ M Nitroblue Tetrazolium (NBT) and 16 μ M Phenazin Methosulfate (PMS). Polysaccharides were added to the reaction buffer to the final concentrations of 2.5, 5, 10, 20, 40, 80 and 160 μ g mL⁻¹ and absorbencies at 517 nm after 30 min reaction were recorded. For the control, NADH was substituted with Tris-HCl buffer.

Measurement of the hydroxyl radical scavenging activity: The measurement of hydroxyl radical scavenging activity was following the protocol described by Smirnov and Cumbe (1989). The reaction solution contained 150 mM sodium phosphate (pH 7.4), 0.15 mM FeSO₄-EDTA, 2 mM sodium salicylate and 6 mM H₂O₂. Different polysaccharides were added to the final concentrations of 0.5-400 μ g mL⁻¹. For the control, H₂O₂ was replaced with sodium phosphate buffer (150 mM, pH 7.4). The mixtures were incubated at 37°C for 1 h and the absorbencies at 510 nm were recorded.

Measurement of the •DPPH radical scavenging activity: Measurement of the •DPPH radical scavenging activity was following the protocol described by Zhang *et al.* (2003). Free radicals were generated by 1,1-Diphenyl-Picryl-Hydrazil (DPPH). Briefly, DPPH was dissolved in minimal toluene and mixed with 50% ethanol to a final concentration of 127 μ mol L⁻¹. The reaction system (2 mL) contained 1.9 mL DPPH and different concentrations (15-250 μ g mL⁻¹) of polysaccharides. The absorbances of the mixtures were measured with a spectrophotometer at 525 nm after 20 min.

Measurement of the red cell hemolysis induced by H₂O₂ scavenging: Scavenging activities of polysaccharides on the rat erythrocytes hemolysis induced by H₂O₂ were measured according to the protocol described by Li *et al.* (2000). In brief, heparin-sodium was added to the rat

blood as the anticoagulant and the erythrocytes were separated from the rat blood by centrifugation and resuspended in 1% physiological saline. The reaction solution (3 mL) contained 1 mL erythrocyte suspension, different concentrations (2.5-200 μ g mL⁻¹) of polysaccharides and 1 mL 400 μ M L⁻¹ H₂O₂. The mixture was incubated at 37°C for 1 h followed by addition of six volumes of physiological saline and centrifugation at 3000 rpm for 6 min. The absorbance of the supernatant was then measured with a spectrophotometer at 415 nm. For the control, no polysaccharide was added.

Measurement of the lipid peroxide inhibition activity in the rat liver homogenate: The rat liver homogenate was prepared following the protocol described by Li *et al.* (2000). Immediately after the mice were sacrificed, the liver tissues were separated and suspended in cold 0.5% physiological saline. The suspended solution was then centrifuged at 500 rpm at 4°C for 10 min to obtain the supernatant. To measure the inhibition effects of polysaccharides on the lipid peroxide in mice liver homogenate, the supernatant (1 mL) was incubated with 1 mL polysaccharide (2-280 μ g mL⁻¹), 100 μ L 6 mM FeSO₄ and 40 μ L 20 mM H₂O₂ at 37°C for 1 h. The reaction was stopped by 15% trichloroacetic acid (1 mL), followed by 0.7% 2-thiobarbituric acid (1 mL). After the protein precipitate was removed by centrifugation at 1000 rpm and 10 min, the absorbance at 532 nm was recorded. The color formation is due to the oxidation product of lipid peroxide, the malondialdehyde-thiobarbituric acid complex. For the control samples, FeSO₄ and H₂O₂ were omitted.

Calculation of the scavenging rate: The percentage of scavenging activity was evaluated according to the following equation, where A stands for the absorbance of the free radical generation system, A_x stands for the absorbance of samples and A₀ stands for the absorbance of the control.

$$\text{Scavenging rate (\%)} = (A - A_x) / (A - A_0) \times 100\%$$

Statistical analysis: Data were processed with the SPSS software (SPSS Inc., IL, USA) and analyzed with the student's t-test.

RESULTS AND DISCUSSION

Effects of the polysaccharide concentration on the scavenging rate for the superoxide radical (O₂⁻): Superoxide radicals were generated by the NADH-PMS-NBT system. The superoxide radical scavenging rates

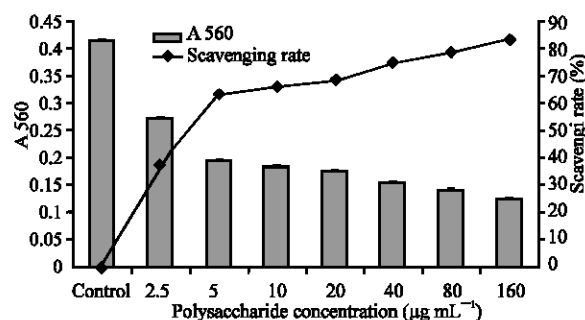


Fig. 1: Effects of the Porphyra polysaccharide concentration on O_2^- scavenging (Mean \pm SD, n=6) (p<0.01)

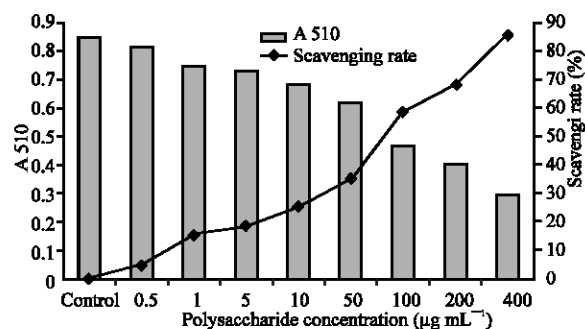


Fig. 2: Scavenging rates for $\bullet OH$ with the Porphyra polysaccharide (Mean \pm SD, n=6) (p<0.01)

with different concentrations of Porphyra polysaccharide were summarized in Fig. 1. As shown, the IC_{50} of Porphyra polysaccharide was below $5 \mu g mL^{-1}$, indicating that the superoxide radical scavenging activity of polysaccharides was significant. The scavenging rate reached 82.77% when the polysaccharide concentration was $160 \mu g mL^{-1}$. Superoxide radicals were generated by the PMS-NADH-NBT system. Polysaccharides were added to the reaction buffer (3mL) to the final concentrations of 2.5, 5, 10, 20, 40, 80 and $160 \mu g mL^{-1}$ and absorbencies at 517 nm after 30 min reaction were recorded. For the control, NADH was substituted with Tris-HCl buffer

Effects of the polysaccharide concentration on the scavenging rate for hydroxyl free radicals: The effects of the polysaccharide concentration on the hydroxyl radical scavenging rate were summarized in Fig. 2. As shown, Porphyra polysaccharide also possessed hydroxyl radical scavenging activity. The scavenging efficiency reached 85.63% when the polysaccharide concentration was $400 g mL^{-1}$. Different polysaccharides were added in the reaction solution to the final concentrations of 0.5-400 $\mu g mL^{-1}$. For the control, H_2O_2 was replaced with sodium

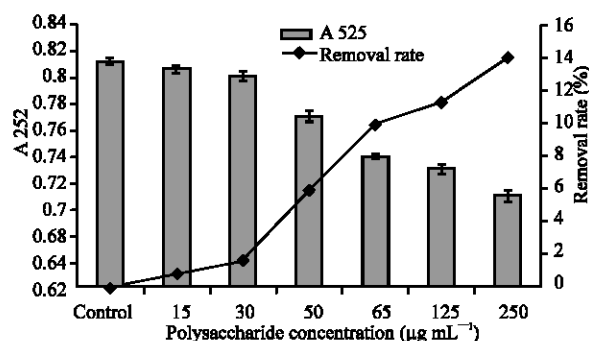


Fig. 3: The Porphyra polysaccharide on the removal rate of $\bullet DPPH$ (Mean \pm SD, n=6) (p<0.01)

phosphate buffer (150 mM, pH 7.4). The mixtures were incubated at $37^\circ C$ for 1 h and the absorbencies at 510 nm were recorded.

Scavenging effects of the polysaccharide on $\bullet DPPH$ radical:

Figure 3 demonstrated the ability of Porphyra polysaccharide in quenching $\bullet DPPH$ radicals. Compared with the superoxide and the hydroxyl radicals, the scavenging activity of $\bullet DPPH$ radicals was much weaker. Only when the polysaccharide concentration was higher than $125 g mL^{-1}$, was the scavenging efficiency able to reach higher than 10% (Fig. 3). Free radicals were generated by 1,1-diphenyl-picryl-hydrazil (DPPH). DPPH was dissolved in minimal toluene and mixed with 50% ethanol to a final concentration of $127 mol L^{-1}$. The reaction system (2 mL) contained 1.9 mL DPPH and different concentrations ($15-250 \mu g mL^{-1}$) of polysaccharides. The absorbencies of the mixtures were measured with a spectrophotometer at 525 nm after 20 min.

Effects of the polysaccharide on the oxymatic hemolysis of erythrocytes:

The effects of the Porphyra polysaccharide on the hemolysis of the rat erythrocytes induced by H_2O_2 were summarized in Fig. 4. As shown, at higher concentrations, Porphyra polysaccharide exhibited strong protective effects against hemolysis. The IC_{50} of Porphyra polysaccharide on hemolysis of erythrocytes induced H_2O_2 was around $26 g mL^{-1}$. The scavenging efficiency of Porphyra polysaccharide on hemolysis of erythrocytes reached up to 82.90% at the polysaccharide concentration of $208 g mL^{-1}$. Heparin-natrium was added to the rat blood as the anticoagulant and the erythrocytes were separated from the rat blood by centrifugation and resuspended in 1% physiological saline. The reaction solution (3 mL) contained 1 mL erythrocyte suspension, different concentrations ($2.5-200 \mu g mL^{-1}$) of polysaccharides and 1 mL $400 \mu M L^{-1} H_2O_2$. The mixture was incubated at $37^\circ C$ for 1 h followed by addition of six

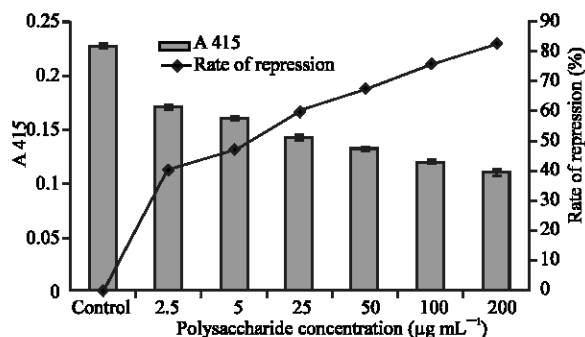


Fig. 4: The inhibitory effects of Porphyra polysaccharide on the oxytatic hemolysis of red blood cells (Mean±SD, n = 6) p<0.01

volumes of physiological saline and centrifugation at 3000 rpm for 6 min. The absorbance of the supernatant was then measured with a spectrophotometer at 415 nm. For the control, no polysaccharide was added.

Effects of the polysaccharide on MDA levels:

Malondialdehyde, formed from the breakdown of polyunsaturated fatty acids, is a convenient indicator for determining the extent of lipid peroxidation. According to our data, the Porphyra polysaccharide fractions from *P. yezoensis* significantly inhibited the microsomal lipid peroxidation (Fig. 5). The IC₅₀ of Porphyra polysaccharide on lipid peroxidation was about 299.07 µg mL⁻¹. The liver tissues were separated from mice and suspended in cold 0.5% physiological saline and then centrifuged at 500 rpm at 4°C for 10 min to obtain the supernatant. The supernatant (1 mL) was incubated with 1 mL polysaccharide (2-280 µg mL⁻¹), 100 µL 6 mmol FeSO₄ and 40 µL 20 mmol H₂O₂ at 37°C for 1 h. The reaction was stopped by 15% trichloroacetic acid (1 mL), followed by 0.7% 2-thiobarbituric acid (1 mL). After the protein precipitate was removed by centrifugation at 1000 rpm for 10 min, the absorbance at 532 nm was recorded. For the control samples, FeSO₄ and H₂O₂ were omitted.

Superoxide anion, hydroxyl and •DPPH free radicals are hazardous to living organisms and they directly associate with many diseases, such as tumor and inflammation. Hence, it is of high importance for the living organisms to get rid of or balance these harmful free radicals (Xin and Liu, 2000). Researches in recent years indicated that polysaccharides from alga possess remarkable anti-oxidation activity. For example, *Sargassum thunbergii* polysaccharide scavenges Reactive Oxygen Species (ROS) effectively with an IC₅₀ of 0.5 mg, as measured by the Py-rugalb Huminol luminescent system (Zhang and Yu, 1997), *Laminaria japonica* fucoidan (IC₅₀ = 20.3 µg mL⁻¹) could scavenge superoxide anion

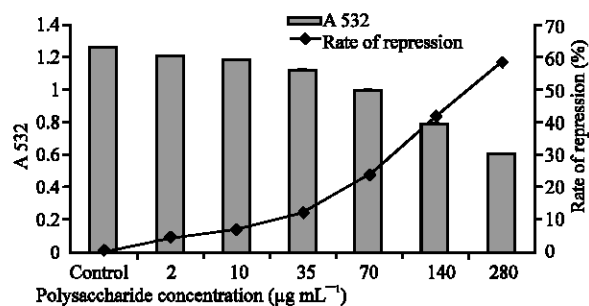


Fig. 5: The inhibitory effects of Porphyra polysaccharide on the MDA levels (Mean±SD, n = 6) (p<0.01)

and to a less extent, the hydroxyl radical and the •DPPH radical (Zhang *et al.*, 2003). For the higher plants, the superoxide anion scavenging ability is generally better. The highest reported values are from the peel extract of the red pomegranate and the extract of *Quercus aliena* acorn, with the IC₅₀ values being 4.01 µg mL⁻¹ and 4.92 µg mL⁻¹ (Jin *et al.*, 2005; Guo *et al.*, 2007). The Porphyra polysaccharides used in our investigation was found to have even stronger scavenging effect on the superoxide radical with an IC₅₀ value of 2.5-5.0 µg mL⁻¹ and scavenging rate of up to 82.77%. The removal efficiencies for •OH and •DPPH were even better and the maximum removal rates were 85.63 and 13.97%, respectively.

The erythrocytes are in an oxygen-fully-loaded environment *in vivo*, with abundant polyunsaturated fatty acids around. The metal-chelating emoglobin around the erythrocytes could easily catalyze lipid peroxidation, thus making erythrocytes susceptible to oxidation injury. Researches on erythrocytes structure and function have provided valuable information for preventing the cellular oxidation injury (Zhang and Yu, 1997; Yang *et al.*, 1999). It has been reported that certain polysaccharide from the alga, e.g., *Laminaria* polysaccharide, possessed protecting roles on the erythrocytes oxidation hemolysis induced by H₂O₂ (Zhang *et al.*, 2003; Luo *et al.*, 2004). Present data suggested that Porphyra polysaccharide effectively inhibited oxytatic hemolysis of erythrocytes induced by H₂O₂. The IC₅₀ value was 25 µg mL⁻¹ and the inhibition rate reached up to 82.90% at the polysaccharide concentration of 200 µg mL⁻¹.

The MDA level increases when the immune system of the organism is down. The redundant lipid peroxides damage the cell membrane and accelerate aging and cell death. Therefore, the MDA level is an important indicator for lipid oxidation (Zhang *et al.*, 2003; Luo *et al.*, 2004; Han *et al.*, 2006). According to present data, at high concentrations, Porphyra polysaccharide inhibited the

increase of MDA levels induced by Fe^{2+} and H_2O_2 in the homogenate of rat liver tissues, indicating that Porphyr polysaccharide could hinder the lipid peroxidation process *in vitro*.

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